

# Acid Base Titration Lab Report Discussion

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## *Experiment 7: ACID-BASE TITRATION: STANDARDIZATION ...*

In this experiment an acid-base titration will be used to determine the molar concentration of a sodium hydroxide (NaOH) solution. Acid-base titrations are also called neutralization titrations because the acid reacts with the base to produce salt and water. During an acid-base titration, there is a point when the number of moles of acid ( $H^+$  ions)

## **Electrical Conductivity of Aqueous Solutions - Colby College**

INTRODUCTION: In this lab you will explore the nature of aqueous solutions by investigating the relationship between conductivity and strong and weak electrolytes. To do this, you will add increasing amounts of either acid or base to several electrolyte solutions. After each addition you will measure the conductivity of the solution.

## *Experiment 2: Acid / base titration - Purdue University*

Titration of the unknown The titration results using standardized NaOH solution are listed in Table 2.

Trial	1*	2	3
Initial volume [mL]	16.60	0.60	16.40
Final volume [mL]	32.30	16.40	32.18
Volume added end-point [mL]	VNaOH	15.70	15.80
	15.78	Table 2. Volume data from the titration of unknown monoprotic acid using standardized	