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Oswaal CBSE Term 2 English Core, Physics, Chemistry & Mathematics Class 12 Sample Question Papers (Set of 4 Books) (Now Based On The CBSE Term-2 Subjective Sample Paper Of Dt. 14 Jan 2022) Oswaal Editorial Board 2022-02-15 Oswaal CBSE Term 2 Sample Paper Class 12 English Core, Physics, Chemistry & Mathematics 2022 Includes 15 Sample Papers. 5 solved & 10 Self-Assessment Papers for Term 2 Board Exams March-April 2022 The CBSE Term 2 Sample Paper Class 12 English Core, Physics, Chemistry & Mathematics 2022 Include all latest typologies of Questions as specified in the latest CBSE Board Sample Paper for Term 2 Board Exams Released on 14th January 2022 These CBSE Term 2 Books Class 12 English Core, Physics, Chemistry & Mathematics 2022 Comprise On-Tips Notes & Revision Notes for Quick Revision Oswaal CBSE Term 2 Sample Papers Class 12 English Core, Physics, Chemistry & Mathematics 2022 Include Mind Maps For Better Learning These CBSE Term 2 Sample Papers Class 12 English Core, Physics, Chemistry & Mathematics 2022 | CBSE Term 2 Books Class 12 English Core, Physics, Chemistry & Mathematics 2022 Help to Prepare Better for Term 2 Board Exams 2022 Get Free E-Assessments of Oswaal360 based on the latest Typologies of Questions as per CBSE Term-II syllabus

Comprehensive Practical Chemistry XI Dr. N. K. Verma 2010-02

Practical/Laboratory Manual Chemistry Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal Dr. S. C. Rastogi 2020-06-23 An Excellent Book in Accordance with the latest syllabus for Class-11 Prescribed by CBSE/NCERT and Adopted by Various State Education Boards. (A) Basic Laboratory Techniques – 1. To cut a glass tube or glass rod, 2. To bend the glass rod at an angle, 3. To draw a glass jet from a glass tube, 4. To bore a cork and fit a glass tube into it. (B) Characterisation and Purification of Chemical Substances- 1. To determine the melting point of the given unknown organic compound and its

identification (simple laboratory technique), 2. To determine the boiling point of a given liquid when available in small quantity (simple laboratory method), 3. To prepare crystals of pure potash alum $[K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O]$ from the given impure sample, 4. To prepare the pure crystals of copper sulphate from the given crude sample, 5. To prepare pure crystals of benzoic acid from a given impure sample. (C) Measurement of pH Values 1. To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper, 2. To determine and compare the pH values of solutions of strong acid (HCl) and weak acid (CH₃COOH) of same concentration, 3. To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper, 4. To study the pH change by common ion (CH₃COO⁻ ion) in case of weak acid (CH₃COOH), 5. To determine the change in pH value of weak base (NH₄OH) in presence of a common ion (NH₄⁺), (D) Chemical Equilibrium 1. To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions, 2. To study the shift in equilibrium between $[Co(H_2O)_6]^{2+}$ and Cl⁻ ions by changing the concentrations of either of the ions, (E) Quantitative Analysis 1. To prepare M/10 oxalic acid solution by direct weighing method, 2. To prepare M/10 solution of sodium carbonate by direct weighing method, 3. To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid, 4. To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution, (F) Qualitative Analysis 1. Analysis of Anions, 2. Analysis of Cations (G) Detection of Elements in Organic Compounds 1. To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne's test, 2. To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number by Lassaigne's test INVESTIGATORY PROJECTS (A) Checking of Bacterial Contamination in Water 1. To check the bacterial contamination in drinking water by testing sulphide ions (B) Methods of Water Purification 1. To purify water from suspended impurities by using sedimentation, 2. To purify water by boiling, 3. To purify water by distillation method, 4. To purify water by reverse osmosis technique. 5. To purify water by GAC method, 6. To purify water by bleach treatment, 7. To purify water by oxidising agent, 8. To purify water by ozone treatment method. (C) Water Analysis 1. To test the hardness of different water samples. (D) Foaming Capacity of Various Soaps 1. To compare the foaming capacity of different washing soaps, 2. To study the effect of addition of sodium carbonate on foaming capacity of washing soap (E) Tea Analysis 1. To study the acidity of different samples of tea leaves (tea) by using pH paper (F) Analysis of Fruits and Vegetable Juices 1. To analyse the fruit and vegetable juices for the constituent present in them (G) Rate of Evaporation 1. To study the rate of evaporation of different liquids (H) Effect of Acids and Bases on Tensile Strength of Fibres 1. To compare the tensile strength of natural fibres and synthetic fibres, 2. To study the effect of acids and bases on tensile strength of different fibres. Log & Antilog Table

This product covers the following: Strictly as per the Full syllabus for Board 2022-23 Exams Includes Questions of the both - Objective & Subjective Types Questions Chapterwise and Topicwise Revision Notes for in-depth study Modified & Empowered Mind Maps & Mnemonics for quick learning Concept videos for blended learning Previous Years' Board Examination Questions and Marking scheme Answers with detailed explanation to facilitate exam-oriented preparation. Examiners comments & Answering Tips to aid in exam preparation. Includes Topics found Difficult & Suggestions for students. Includes Academically important Questions (AI) Dynamic QR code to keep the students updated for 2023 Exam paper or any further ISC notifications/circulars

Hard Bound Lab Manual Chemistry Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar Lab Manuals

Oswaal CBSE Question Bank Class 12 English, Physics, Chemistry & Mathematics (Set of 4 Books) (For 2022-23 Exam) Oswaal Editorial Board 2022-05-26 Oswaal CBSE Question Bank Class 12 Physics, Chemistry & Mathematics 2022-23 are based on latest & full syllabus The CBSE Question Bank Class 12 Physics, Chemistry & Mathematics 2022-23 Includes Term 1 Exam paper 2021+Term II CBSE Sample paper+ Latest Topper Answers The CBSE Books Class 12 2022 -23 comprises Revision Notes: Chapter wise & Topic wise The CBSE Question Bank Class 12 Physics, Chemistry & Mathematics 2022-23 includes Exam Questions: Includes Previous Years Board Examination questions (2013-2021) It includes CBSE Marking Scheme Answers: Previous Years' Board Marking scheme answers (2013-2020) The CBSE Books Class 12 2022 -23 also includes New Typology of Questions: MCQs, assertion-reason, VSA ,SA & LA including case based questions The CBSE Question Bank Class 12 Physics, Chemistry & Mathematics 2022-23 includes Toppers Answers: Latest Toppers' handwritten answers sheets Exam Oriented Prep Tools Commonly Made Errors & Answering Tips to avoid errors and score improvement Mind Maps for quick learning Concept Videos for blended learning The CBSE Question Bank Class 12 Physics, Chemistry & Mathematics 2022-23 includes Academically Important (AI) look out for highly expected questions for the upcoming exams

Practical/Laboratory Manual Chemistry Class XII based on NCERT guidelines by Dr. S. C. Rastogi, Er. Meera Goyal Dr. S. C. Rastogi 2020-06-22 A. Surface Chemistry 1.To prepare colloidal solution (sol) of starch, 2. To prepare a colloidal solution of egg albumin 3.To prepare colloidal solution of gum, 4. To prepare colloidal solution of aluminium hydroxide $[Al(OH)_3]$, 5.To prepare colloidal solution of ferric hydroxide $[Fe(OH)_3]$, 6.To prepare colloidal solution of arsenious sulphide $[As_2S_3]$, 7. To purify a freshly prepared sol by dialysis, 8. To compare the effectiveness of different common oils (Castor oil, cotton seed oil, coconut oil, kerosene oil, mustard oil) in forming emulsions. Viva-Voce B. Chemical Kinetics 1. To study the effect of concentration on the rate of reaction between sodium thiosulphate and hydrochloric acid, 2. To study the effect of temperature on the rate of reaction between sodium thiosulphate and hydrochloric acid, 3. To study the rate of reaction of iodide ions with hydrogen peroxide at different concentrations of iodide ions, 4. To study the

rate of reaction between potassium iodate (KIO_3) and sodium sulphite (Na_2SO_3) using starch solution as indicator

Viva-Voce C. Thermochemistry

1. Determine the enthalpy of dissolution of copper sulphate ($CuSO_4 \cdot 5H_2O$) in water at Room temperature,
2. To determine the enthalpy of neutralization of the reaction between HCl and $NaOH$,
3. To determine enthalpy change during the interaction between acetone and chloroform

Viva-Voce D. Electrochemistry

1. To study the variation of cell potential in $Zn|Zn^{2+}||Cu^{2+}|Cu$, with change in concentration of electrolytes ($CuSO_4$ or $ZnSO_4$) at room temperature

Viva-Voce E. Chromatography

1. To separate the coloured components (pigment) present in the given extract of leaves and flowers by ascending paper chromatography and find their R_f values,
2. To separate the coloured components present in the mixture of red and blue inks by ascending paper chromatography and find their R_f values,
3. To separate Co^{2+} and Ni^{2+} ions present in the given mixture by using ascending paper chromatography and determine their R_f values

Viva-Voce F. Preparation of Inorganic Compounds

1. Preparation of double salt of ferrous ammonium sulphate (Mohr's salt) from ferrous sulphate and ammonium sulphate,
2. To prepare a pure sample of potash alum (fitkari),
3. Preparation of crystals of potassium ferric oxalate or potassium trioxalato ferrate (III)

Viva-Voce G. Preparation of Organic Compounds

1. Preparation of iodoform from ethyl alcohol or acetone,
2. Preparation of acetanilide in laboratory,
3. Preparation of *p*-Naphthol aniline dye,
4. To prepare a pure sample of dibenzalacetone,
5. To prepare a pure sample of *p*-nitro acetanilide

Viva-Voce H. Tests for the Functional Groups Present in Organic Compounds

Viva-Voce I. Study of Carbohydrates, Fats and Proteins

1. To study simple reactions of carbohydrate,
2. To study simple reactions of fats,
3. To study simple reactions of proteins,
4. To investigate presence of carbohydrates, fats and proteins in food stuffs

Viva-Voce J. Volumetric Analysis

1. To prepare 250 ml of $M/10$ solution of oxalic acid,
2. To prepare 250 ml of $M/10$ solution of ferrous ammonium sulphate,
3. Prepare $M/20$ solution of oxalic acid, with its help find out the molarity and strength of the given solution of potassium permanganate,
4. Prepare $M/20$ solution of Mohr's salt, using this solution determine the molarity and strength of potassium permanganate solution

Viva-Voce K. Qualitative Analysis

INVESTIGATORY PROJECTS

1. To study the presence of oxalate ions in guava fruit at different stages of ripening.
2. To study the quantity of caseine present in different samples of milk.
3. Preparation of soyabean milk and its comparison with natural milk with respect to curd formation, effect of temperature etc.
4. To study the effect of potassium bisulphite as food preservative at various concentrations.
5. To study the digestion of starch by salivary amylase and the effect of pH and temperature on it.
6. To study and compare the rate of fermentation of the following materials—wheat flour, gram flour, potato juice and carrot juice.
7. To extract essential oils present in saunf (aniseed), ajwain (corum), illaichi (cardomom).
8. To detect the presence of adulteration in fat, oil and butter,
9. To investigate the presence of NO_2^- in brinjal.

Chemistry Lab Manual Class XI | follows the latest CBSE syllabus and other State Board following the CBSE Curriculum. Mr. Rohit Manglik 2022-08-04 With the NEP 2020 and expansion of research and knowledge has changed the face of education to a great extent. In the Modern times, education is not just

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constricted to the lecture method but also includes a practical knowledge of certain subjects. This way of education helps a student to grasp the basic concepts and principles. Thus, trying to break the stereotype that subjects like Physics, Chemistry and Biology means studying lengthy formulas, complex structures, and handling complicated instruments, we are trying to make education easy, fun, and enjoyable.

Oswaal CBSE One for All Class 12 English, Physics, Chemistry & Mathematics (Set of 4 books) (For 2023 Exam) Oswaal Editorial Board 2022-09-12 Chapter Navigation Tools CBSE Syllabus : Strictly as per the latest CBSE Syllabus dated: April 21, 2022 Cir. No. Acad-48/2022 Latest updations: 1. Term I & Term II Solved Papers 2022-23 (all sets of Delhi & Outside Delhi) 2. Toppers Answers -2020 Revision Notes: Chapter wise & Topic wise Exam Questions: Includes Previous Years Board Examination questions (2013-2021) CBSE Marking Scheme Answers: Previous Years' Board Marking scheme answers (2013-2020) with detailed explanation to facilitate exam-oriented preparation. New Typology of Questions: MCQs, assertion-reason, VSA, SA & LA including case based questions Toppers Answers: Latest Toppers' handwritten answers sheets Questions from Board Question Bank -2021 Mind Maps and concept videos to make learning simple. Coverage of Chapter wise complete NCERT textbook + NCERT Exemplar questions with answers. Dynamic QR code to keep the students updated for any further CBSE notifications/circulars Commonly Made Errors & Answering Tips to avoid errors and score improvement Self Assessment Tests & Practice Papers for self - evaluation

Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal 2021-05-29

1. Basic Laboratory Techniques 1. To cut a glass tube or glass rod, 2. To bend the glass rod at an angle, 3. To draw a glass jet from a glass tube 4. To bore a cork and fit a glass tube into it Viva-Voce 2. Characterisation and Purification of Chemical Substances 1. To determine the melting point of the given unknown organic compound and its identification (simple laboratory technique) Viva-Voce 2. To determine the boiling point of a given liquid when available in small quantity (simple laboratory method) Viva-Voce 3. To prepare crystals of pure potash alum $[K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O]$ from the given impure sample 4. To prepare the pure crystals of copper sulphate from the given crude sample 5. To prepare pure crystals of benzoic acid from a given impure sample Viva-Voce 3. Measurement of pH Values 1. To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper 2. To determine and compare the pH values of solutions of strong acid (HCl) and weak acid (CH₃COOH) of same concentration 3. To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper 4. To study the pH change by common ion (CH₃COO⁻ ion) in case of weak acid (CH₃COOH) 5. To determine the change in pH value of weak base (NH₄OH) in presence of a common ion (NH₄⁺) Viva-Voce 4. Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2. To study the shift in equilibrium between $[Co(H_2O)_6]^{2+}$ and Cl⁻ ions by changing the concentrations of either of the ions Viva-Voce 5. Quantitative Analysis 1. To prepare M/10 oxalic acid solution by direct weighing

method 2. To prepare M/10 solution of sodium carbonate by direct weighing method
 3. To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid
 4. To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution Viva-Voce
 6. Qualitative Analysis Analysis of Anions Analysis of Cations Viva-Voce
 7. Detection of Elements in Organic Compounds
 1. To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne's test
 2. To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number by Lassaigne's test Viva-Voce
 INVESTIGATORY PROJECTS
 1. Checking of Bacterial Contamination in Water
 1. To check the bacterial contamination in drinking water by testing sulphide ions Viva-Voce
 2. Methods of Water Purification
 1. To purify water from suspended impurities by using sedimentation
 2. To purify water by boiling
 3. To purify water by distillation method
 4. To purify water by reverse osmosis technique
 5. To purify water by GAC method
 6. To purify water by bleach treatment
 7. To purify water by oxidising agent
 8. To purify water by ozone treatment method Viva-Voce
 3. Water Analysis
 1. To test the hardness of different water samples Viva-Voce
 4. Foaming Capacity of Various Soaps
 1. To compare the foaming capacity of different washing soaps
 2. To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva-Voce
 5. Tea Analysis
 1. To study the acidity of different samples of tea leaves (tea) by using pH paper Viva-Voce
 6. Analysis of Fruits and Vegetable Juices
 1. To analysis the fruit and vegetable juices for the constituent present in them Viva-Voce
 7. Rate of Evaporation
 1. To study the rate of evaporation of different liquids lViva-Voce
 8. Effect of Acids and Bases on Tensile Strength of Fibres
 1. To compare the tensile strength of natural fibres and synthetic fibres
 2. To study the effect of acids and bases on tensile strength of different fibres Viva-Voce

CBSE MCQs Chapterwise For Term I & II, Class 12, Chemistry Career Point Kota 2021-10-12 Strictly as per the new term-wise syllabus for Board Examinations to be held in the academic session 2021-22 for class 12. Multiple Choice Questions based on new typologies introduced by the board- Stand-Alone MCQs, MCQs based on Assertion-Reason, Case-based MCQs. Include Questions from CBSE official Question Bank released in April 2021 Answer key with Explanations Sample Paper on the latest pattern of Term - 1 exam.

Saraswati Chemistry Class 09 RP Manchanda A text book on Chemistry

Oswaal CBSE Chapterwise & Topicwise Question Bank Class 11 Chemistry Book (For 2022-23 Exam) Oswaal Editorial Board 2022-08-12 Chapter Navigation Tools • CBSE Syllabus : Strictly as per the latest CBSE Syllabus dated: April 21, 2022 Cir. No. Acad-48/2022 Latest Updatons: 1. All new topics/concepts/chapters were included as per the latest curriculum. 2. Self Assessment papers for practice • Revision Notes: Chapter wise & Topic wise • Exam Questions: Includes Previous Years KVS exam questions • New Typology of Questions: MCQs, VSA, SA & LA including case based questions • NCERT Corner: Fully Solved Textbook Questions (Exemplar Questions in Physics, Chemistry, Biology) Exam Oriented Prep Tools •

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Unit wise test for self preparation

Teaching Chemistry Around the World Björn Risch 2010 As teachers we often tend to expect other countries to teach chemistry in much the same way as we do, but educational systems differ widely. At Bielefeld University we started a project to analyse the approach to chemical education in different countries from all over the world: Teaching Chemistry around the World. 25 countries have participated in the project. The resulting country studies are presented in this book. This book may be seen as a contribution to make the structure of chemistry teaching in numerous countries more transparent and to facilitate communication between these countries. Especially in the case of the school subject chemistry, which is very unpopular on the one hand and occupies an exceptional position on the other hand – due to its relevance to jobs and everyday life and most notably due to its importance for innovation capacity and problem solving – we have to learn from each others' educational systems.

Investigatory Projects in Chemistry Pradip Tekade 2011-07 In recent years, the courses in chemistry have been considered extremely important for students desirous of pursuing basic science as well as technical education. Taking into consideration this trend, this book has been specially written for students who are interested in investigatory and innovative projects in chemistry. The unique feature of this book is that the basic, theory, procedure and conclusions for each project are given in comprehensive manner. In all 51 projects are included in this book from all the branches of chemistry viz. inorganic, organic, physical, analytical and general chemistry. Some of them are: Antacid effectiveness analysis, Study of tea chemistry, study the foaming capacity of soaps, Study of constituents of brass and bronze, Preparation of potash alum from scrap aluminum, Extraction of essential oil from aniseed, preparation of pigments and poster paints using chemicals and reagents, Analysis of calcium, phosphate, chloride, magnesium and iron from bone ash, To Study the Setting of Cement, Comparative study and qualitative analysis of cold drinks, preparation of cuprammonium rayon threads from filter paper, Analysis of talcum powder etc.

Methods Of Teaching Chemistry K.S. Kumar 2004 Contents: Introduction, Scope and Influence, Past Experience, Objectives and Aims, Teaching under Scheme, Methods of Teaching, Role of Teacher, Measurement and Evolution, Curriculum Development, Broadbased Curriculum, Enrichment of Controls, Planning the Lesson, Teaching Devices, Audio-Visual Aids, Role of Laboratory, A Rich Laboratory, New Trends, Place among other Discipline.

Comprehensive Experimental Chemistry V. K. Ahluwalia 1997 This Book Has Been Especially Written For Class Xii Students Under 10+2 Pattern Of Education According To The Syllabi Prescribed By The Cbse And Other States Boards. This

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Book Will Help The Students In Acquiring Correct Skills In Practicals And Various Techniques Of All Laboratory Experiments. Salient Features * An Introduction To The Book Is Given. This Describes The Laboratory Apparatus And Instructions And Precautions For Working In The Laboratory. * Simple Language And Lucid Style. * Adequate Number Of Illustrations To Explain And To Clarify The Use Of Various Apparatus Used In The Laboratory. * Theoretical Aspects Of Each Equipment Have Been Discussed Along With Experiments. * In Volumetric Analysis, Both The Normality And Molarity Concepts Are Made Clear. * In Quantitative Analysis (Inorganic And Organic), Various Tests Have Been Given In A Systematic Way. Specimen Recordings Of Experiments Are Given To Help The Students To Record On Their Notebooks. * Viva-Voice Questions Have Been Included In Each Chapter. * A Fairly Large Number Of Investigatory Projects Covering Various Topics Are Given. Selection Of Projects Is Carefully Made Which Can Be Easily Performed In School Laboratory. * An Appendix Describing Various Chemical Hobbies Is Given Which Will Be Extremely Helpful To The Students For The Development Of Chemical Hobbies, Understanding The Basic Principles Involved And The Chemistry Of Various Hobbies. * An Appendix Describing Some Typical Chemical Exhibits Is Also Given. This Will Help The Students To Participate In The Science Fairs Organized By Various Agencies. These Experiments Will Cultivate Interest Among The Students For Learning Chemistry. * An Appendix Each For The Solubility'S Of Various Salts, Atomic Weights, Preparation Of Various Reagents, Indicator Papers And The First Aid To Be Administered In Case Of Accidents Is Given. The Syllabi Prescribed For Class Xii Students Under 10+2 Pattern Along With Distribution Of Marks Is Also Given.

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Methods Of Teaching Science K.Jaya Sree 2010 The method of teaching each subject play a pivotal role in enhancing the efficiency of their practitioners. Identifying the very importance of the methods of teaching and the quality of books, a series of books on the methods of teaching different subjects have been developed by experienced teacher educators for the benefit of teachers in making in teacher education institutions. Contents: Teacher s Role, Teaching Techniques, Methods of Vogue, Approaches in Vogue, Aims and Objectives of Teaching, Advancement of Science in India, Behaviour and Objectives, Educational Technology, Audio-visual Aids in Use, Experiments in Innovation, Programmes for Enrichment, Instruction in a Programmed Manner, Individual Level Instructions, Planning the Lessons, Curriculum (India), Curriculum (World), Textbook and Material Projects, Social Service.

Chemistry R P Manchanda A text book on Chemistry

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Comprehensive Practical Chemistry XII Dr. N . K. Verma 2011-11-01

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Lab Manual Chemistry Class XII -by Dr. K. N. Sharma, Dr. Subhash Chandra Rastogi, Er. Meera Goyal (SBPD Publications) Dr. K. N. Sharma 2021-07-03 Highly Useful for Various Engineering and Medical Competitive Examinations.

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Career Point Kota Career Point Kota 2020-07-28 The CBSE has made certain changes in the assessment structure from the session 2019-20 onwards. In the new scheme of examination, CCE and term system has been replaced with the Internal Assessment & Single Annual Exam by CBSE itself. Single exam conducted by CBSE will carry 80 marks whereas 20 marks are left to the schools for internal assessment. CBSE has issued detailed guidelines on how the internal marks will be divided among different activities. From 2019 onwards, there will be internal choices in board examinations with increased internal options in the question paper. Considering this change, now a student has to prepare accordingly for board examinations. The new assessment format brought with it excitement as well as anxiety. And to help them prepare and excel in their CBSE board examination, Career Point Kota has developed a series of 10 Most Likely Question Papers with Solutions. The Key Features of Most Likely Question Papers with Solutions Series : New OBJECTIVE TYPE question in each paper. Syllabus of CBSE 2019-20. Based on the latest CBSE Syllabus & Pattern. Mind map of each chapter is given to visualize and help acquire a better understanding. Important terms, facts, formulae and quick revision tips are given. Covers questions asked in previous year board exams. Toppers Answer Sheet as released by CBSE to understand the scoring technique. We hope this book will gratify students' need for the new CBSE pattern board exam and smoothen their path to success. We wish to utilize the opportunity to place on record our special thanks to all the members of the Content Development team for their efforts to create this wonderful book.

Chemistry Lab Manual Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar Lab Manual

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