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Chapter Thirteen NUCLEI - National Council of Educational ...

atomic mass unit (u), defined as 1/12th of the mass of the carbon (^{12}C) atom. According to this definition mass of one C atom $12 \text{ u} = 12 \times 1.992647 \times 10^{-27} \text{ kg} = 2.3911764 \times 10^{-26} \text{ kg}$ (13.1) The atomic masses of various elements expressed in atomic mass unit (u) are close to being integral multiples of the mass of a hydrogen atom.

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